Received 16 December 2004 Accepted 8 March 2005

Online 18 March 2005

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# Ray J. Butcher,<sup>a</sup>\* Tushar S. Basu Baul,<sup>b</sup> Keisham Surjit Singh<sup>b</sup> and Frank E. Smith<sup>c</sup>

<sup>a</sup>Department of Chemistry, Howard University, 525 College Street NW, Washington, DC 20059, USA, <sup>b</sup>Department of Chemistry, North-Eastern Hill University, NEHU Permanent Campus, Umshing, Shillong 793 022, India, and <sup>c</sup>Weill Cornell Medical College in Qatar, PO Box 24144, Doha, Qatar

Correspondence e-mail: butcher@harker.nrl.navy.mil

#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.007 Å R factor = 0.076 wR factor = 0.216 Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, 2-{(E)-3-[(E)-4-chlorophenylimino]-4-hydroxyphenyldiazenyl}benzoic acid hemihydrate, C<sub>20</sub>H<sub>14</sub>Cl-N<sub>3</sub>O<sub>3</sub>·0.5H<sub>2</sub>O, forms sheets, parallel to the (100) plane, that are linked by weak aromatic C-H···O interactions. These sheets are in turn linked by strong hydrogen bonds to interstitial water molecules.

an azo and a Schiff base linkage

A quasi-planar polyaromatic compound containing

## Comment

Aromatic monoamines react with 2-[(E)-2-(3-formyl-4hydroxyphenyl)-1-diazenyl]benzoic acid to form products, such as the title compound, (I), which contain both azo and Schiff base (*i.e.* imino) linkages. Recently, a few tributyltin(IV) complexes containing similar azo and Schiff base components were obtained which exhibit interesting biological properties (Basu Baul et al., 2004). Their structures were determined using spectroscopic techniques and single-crystal X-ray crystallography. A similar azo-Schiff base organic compound, namely N-p-methoxybenzylidene-p-phenylazoaniline, is a mesogen and exhibits a nematic liquid-crystal phase; its X-ray structure has also been determined recently (Revannasiddaiah et al., 1997). In addition, structures of two similar molecules containing meta azo and imine substituents, also exhibiting liquid crystalline properties, have recently been determined (Baumeister et al., 1990; Cariati et al., 2004). The potential usefulness of such systems has prompted us to determine the structure of the title compound, (I).



The title compound (Fig. 1) crystallizes in the monoclinic space group C2/c with Z = 8. Interstitial water molecules (one for each two of the title molecules) are located on four twofold axes that pass through the unit cell. The three-ring system assumes an extended conformation, with both outer rings slightly twisted with respect to the central aromatic ring. These quasi-planar molecules form sheets, parallel to the (100) plane (see Fig. 2), that are linked by weak aromatic  $C-H\cdots O$ interactions. Adjacent sheets are linked by strong hydrogen bonds to interstitial water molecules (see Fig. 3). Bond lengths and angles within the molecule are unremarkable and similar to those found in related molecules (Baumeister *et al.*, 1990; Cariati *et al.*, 2004).

Printed in Great Britain - all rights reserved

© 2005 International Union of Crystallography





View of the molecule of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 20% probability level. H atoms have been omitted. Hydrogen bonds are shown as dashed lines.



### Figure 2

The molecular packing of (I), viewed down the a axis, perpendicular to the molecular sheets. Dashed lines indicate hydrogen bonds.

## Experimental

*p*-Chloroaniline in hot ethanol was condensed with with an equimolar amount of 2-[(*E*)-2-(3-formyl-4-hydroxyphenyl]-1-diazenylbenzoic acid (Basu Baul *et al.*, 1996) in hot toluene to yield the title compound, (I) (Basu Baul *et al.*, 2004), in 56% yield. Recrystallization from chloroform and dimethyl sulfoxide solution (3:1) gave orange crystals (m.p. 498–499 K). Analysis found: C 63.20, H 3.65, N 11.13%; calculated for  $C_{20}H_{15}ClN_3O_{3.5}$ : C 63.25, H 3.72, N 11.06%.

Crystal data

$C_{20}H_{14}ClN_{3}O_{3} \cdot 0.5H_{2}O$	$D_x = 1.422 \text{ Mg m}^{-3}$
$M_r = 388.80$	Cu Ka radiation
Monoclinic, C2/c	Cell parameters from 49
a = 13.2451 (19)  Å	reflections
b = 21.668 (4) Å	$\theta = 5.4-27.8^{\circ}$
c = 12.664 (2) Å	$\mu = 2.12 \text{ mm}^{-1}$
$\beta = 91.69(2)^{\circ}$	T = 293 (2)  K
$V = 3632.8 (10) \text{ Å}^3$	Prism, red
Z = 8	$0.50 \times 0.35 \times 0.28 \text{ mm}$
Data collection	
Bruker P4 diffractometer	$R_{\rm int} = 0.037$
$2\theta/\omega$ scans	$\theta_{\rm max} = 69.1^{\circ}$
Absorption correction: refined from	$h = 0 \rightarrow 16$
$\Delta F$ (SHELXA; Sheldrick, 1998)	$k = 0 \rightarrow 26$
$T_{\min} = 0.415, T_{\max} = 0.552$	$l = -15 \rightarrow 15$
3534 measured reflections	3 standard reflections
3382 independent reflections	every 97 reflections

3382 independent reflectionsevery 97 reflections1707 reflections with  $I > 2\sigma(I)$ intensity decay: 0.3%



## Figure 3

The molecular packing of (I), viewed down the b axis, showing how the aromatic sheets are linked by interstitial water molecules. Dashed lines indicate hydrogen bonds.

Refinement

 $\begin{array}{ll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_o^2) + (0.0577P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.076 & + 7.4356P] \\ wR(F^2) = 0.216 & where \ P = (F_o^2 + 2F_c^2)/3 \\ S = 1.06 & (\Delta/\sigma)_{\rm max} < 0.001 \\ 3382 \ {\rm reflections} & \Delta\rho_{\rm max} = 0.20 \ {\rm e} \ {\rm \AA}^{-3} \\ 254 \ {\rm parameters} & \Delta\rho_{\rm min} = -0.25 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$ 

### Table 1

Selected geometric parameters (Å, °).

Cl-C18	1.730 (5)	N1-C7	1.426 (6)
O1-C1	1.207 (6)	N2-C8	1.411 (6)
O2-C1	1.311 (6)	N3-C14	1.287 (6)
O3-C11	1.333 (6)	N3-C15	1.407 (6)
N1-N2	1.269 (5)		
N2-N1-C7	114.2 (4)	C2-C7-N1	117.1 (4)
N1-N2-C8	114.3 (4)	C13-C8-N2	114.9 (4)
C14-N3-C15	123.0 (4)	C9-C8-N2	126.1 (4)
O1-C1-O2	118.6 (5)	N3-C14-C12	120.3 (5)
O1-C1-C2	122.1 (5)	C20-C15-N3	125.5 (4)
O2-C1-C2	119.2 (5)	C16-C15-N3	116.3 (4)
C6-C7-N1	122.6 (4)		
C7-N1-N2-C8	174.3 (4)	C15-N3-C14-C12	-179.2 (4)
N2-N1-C7-C2	168.7 (4)	C13-C12-C14-N3	177.9 (4)
N1-N2-C8-C13	177.0 (4)	C14-N3-C15-C16	-161.4 (5)

Table 2		
Hydrogen-bonding geometry	(Å,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O2−H2O···N1	0.82	1.83	2.580 (5)	152
O3−H3O···N3	0.82	1.81	2.544 (5)	148
$O1W - H1W1 \cdots O1$	0.852 (10)	2.05 (3)	2.875 (5)	164 (8)
$C5-H5A\cdotsO1^{i}$	0.93	2.56	3.457 (7)	161
$C14-H14A\cdots O1W^{ii}$	0.93	2.47	3.364 (7)	162
$C19-H19A\cdots O2^{ii}$	0.93	2.59	3.355 (6)	140

Symmetry codes: (i)  $x, 1 - y, \frac{1}{2} + z$ ; (ii) x, y, 1 + z.

After location of the H atoms in a difference map, their positions were idealized. The H atoms were allowed to ride on their parent C atoms, with C-H = 0.93 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$ . Positional parameters for the water H were refined, but the  $U_{iso}(H)$  value was fixed at  $1.5U_{eq}(O)$ .

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Sheldrick, 1997); program(s) used to solve structure: *SHELXTL* ; program(s) used to refine

structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

RJB acknowledges the DoD HBCU program for funds to upgrade the diffractometer. TSBB acknowledges financial support from the Department of Science & Technology, New Delhi, India (grant No. SP/S1/F26/99, TSBB).

## References

Baumeister, U., Kosturkiewicz, Z., Hartung, H., Demus, D. & Weissflog, W. (1990). Liq. Cryst. 7, 241–249.

- Basu Baul, T. S., Pyke, S. P., Sarma, K. K. & Tiekink, E. R. T. (1996). *Main Group Met. Chem.* **19**, 807–814.
- Basu Baul, T. S., Singh, K. S., Song, X., Zapata, A., Eng, G., Lycka, A. & Linden, A. (2004). J. Organomet. Chem. 689, 4702–4711.
- Cariati, F., Caruso, U., DeMaria, A., Fusco, M., Panunzi, B., Roviello, A. & Tuzi, A. (2004). *Inorg. Chim. Acta*, 357, 548–556.
- Revannasiddaiah, D., Lokanath, N. K., Sridhar, M. A. & Prasad, J. S. (1997). Z. *Kristallogr.* **212**, 387–388.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison Wisconsin, USA.
- Sheldrick, G. M. (1998). SHELXA. Release 97-2. University of Göttingen, Germany.
- Siemens (1994). XSCANS. Version 2.20. Siemens Analytical X-ray Instruments Inc. (now Bruker AXS Inc.), Madison, Wisconsin, USA.